### **Sample Format (Lesson Plan)**

### MCM DAV College for Women, Sector – 36A, Chandigarh Monthly Teaching Plans (Odd Semester) Session – (2021-22)

Name of the Teacher/s: 1. Dr. Shweta Sareen

**Department** <u>Chemistry</u>

# Class\_B.Sc. I Subject Physical Chemistry Section (s) A and B

S.No.	Date (Monthly)		<b>Topics to be Covered</b>	Academic Activity Undertaken*
	From	To	7	
1	(Mor	nthly)	Unit 1: Mathematical Concepts and Evaluation of Analytical Data:  Logarithmic relations, curve sketching, linear graphs and calculation of slopes, differentiation and integration of functions like ex, xn, sin x, log x; maxima and minima, partial differentiation and reciprocity relations. Terms of mean and median, precision and accuracy in chemical analysis, determining accuracy of methods,	•
			improving accuracy of analysis, data treatment for series involving relatively few measurements, linear least squares curve fitting, types of errors, standard deviation.	

2	1 <sup>st</sup> October,	27 <sup>th</sup> October,	Unit-II:Gaseous States:	Lecture method
	2021	2021	Postulates of kinetic theory of gases, deviation from ideal behavior, Van der Waal's equation of state. Critical Phenomena: PV isotherms of real gases, continuity of states, the isotherms of Van der Waal's equation, relationship between critical constants and Van der Waal's constants, the law of corresponding states, reduced equation of state. Molecular Velocities: Root mean square, average and most probable velocities. Qualitative discussion of the Maxwell's distribution of molecular velocities, collision number, mean free path and collision diameter. Liquification of gases (based on Joule-Thomson effect).	
3	28 <sup>rd</sup> October, 2021	18 <sup>th</sup> November, 2021	Unit-III: Chemical Kinetics-1  Chemical kinetics and its scope, rate of a reaction, factors influencing the rate of a reaction- concentration, temperature, pressure, solvent, light, catalyst. Concentration dependence of rates, mathematical characteristics of simple chemical reactions – zero order, first order, second order, pseudo	Lecture Method, Online Sources

			order, half life and mean	
4	19 <sup>th</sup> November, 2021	8 <sup>th</sup> December, 2021	life.  Determination of the order of reaction – differential method, method of integration, method of half life period and isolation method.	Lecture Method
			Unit-IV: Chemical Kinetics-II Theories of Chemical Kinetics: Effect of temperature on rate of reaction, Arrhenius equation, concept of activation energy.	
5.	9 <sup>th</sup> December, 2021	Till exam date	Simple collision theory based on hard sphere model, transition state theory (equilibrium hypothesis). Expression for the rate constant based on equilibrium constant and thermodynamic aspects.  Catalysis and general characteristics of catalytic reactions, Homogeneous catalysis, acid-base catalysis and enzyme catalysis including their mechanisms, Michaelis Menten equation for enzyme catalysis and its mechanism.	
Department	cal Meeting to (		Review the Monthly complesson plans	tion of Syllabus as per
5 <sup>th</sup> October, 2021	The teachers h		he scheduled chapters and top plan	oics as shown in the lesson
Department	al Meeting to (		Review the Monthly complesson plans	etion of Syllabus as per
9 <sup>rd</sup> Nov, 2021	The teachers h	ave completed the	he scheduled chapters and top plan	ics as shown in the lesson
Department	al Meeting to (		Review the Monthly complesson plans	tion of Syllabus as per

7 <sup>th</sup> ,December,	The teachers have completed the scheduled chapters and topics as shown in the lesson				
2021	plan				
Department	Departmental Meeting to Coordinate and Review the Monthly completion of Syllabus as per				
lesson plans					
12 <sup>th</sup> Dec,	The teachers have completed the scheduled chapters and topics as shown in the lesson				
2021	plan				
Departmental Meeting to Coordinate and Review the Monthly completion of Syllabus as per					
lesson plans					

<sup>\*</sup>Any of these – (i) Lecture Method; (ii) PPT; (iii) Online Sources; (iv) Group Discussion; (v) Case Studies etc.

Other Methods adopted by the teacher – Please write the specific teaching method

### **Sample Format (Lesson Plan)**

#### MCM DAV College for Women, Sector – 36A, Chandigarh Monthly Teaching Plans (Even Semester) Session – (2021-22)

Name of the Teacher/s: 1. Dr. Shweta Sareen

**Department** <u>Chemistry</u>

# Class\_B.Sc. I Subject Physical Chemistry Section (s) A and B

S.No.	Date (Monthly)		<b>Topics to be Covered</b>	Academic Activity Undertaken*
	From	To	7	
1	3 -02 - 2022	28 -02-2022	Unit 1: Thermodynamics I	Lecture method, Online sources
			<b>Definition of</b>	
			Thermodynamic Terms:	
			System, surroundings	
			etc. Types of systems,	
			intensive and extensive	
			properties. State and	
			path functions and their differentials.	
			Thermodynamic	
			process. Concept of heat	
			and work.	
			and work.	
			First Law of	
			Thermodynamics:	
			Statement, definition of	
			internal energy and	
			enthalpy, Heat capacity,	
			heat capacities at	
			constant volume and	
			pressure and their	
			relationship. Joule's	
			Law-Joule-Thomson	
			coefficient and inversion	
			temperature.	
			Calculations of w, q, dU	

			& dH for the expansion of ideal gases under isothermal and adiabatic conditions for reversible process.	
2	1 -03-2022	15-03-2022	Unit-II: Thermochemistry  Standard state, standard enthalpy of formation-Hess's Law of constant Heat Summation and its applications. Heat of reaction at constant pressure and at constant volume. Enthalpy of neutralization. Bond dissociation energy and its calculation from thermo-chemical data, temperature dependence of enthalpy. Kirchoff's	Lecture method
3	16-03-2022	16-04-2022	equation.  Unit-III: Colloidal State  Definition of colloids, classification of colloids. Solids in liquids (sols): Properties –kinetic, optical and electrical; stability of colloids, protective action, Hardy-Schulze rules, gold number.  Liquids in liquids (emulsions): Types of emulsions, preparation.  Emulsifier. Liquids in solids (gels): Classification, preparation and properties, inhibition, general applications of colloids.	Lecture Method, Online Sources
4	18-04-2022	Till exams	Unit-IV: Solutions, Dilute Solutions and Colligative Properties:	Lecture Method

	T	
		leal and non-ideal
		lutions, methods of
		pressing
		oncentrations of
		lutions, activity and
	ac	ctivity coefficient.
	D	ilute solution,
	co	olligative properties,
	R	aoult's law, relative
	lo	wering of vapour
	pı	essure, molecular
	w	eight determination.
	0	smosis, law of osmotic
	pı	ressure and its
	m	easurement,
	de	etermination of
	m	olecular weight from
	os	emotic pressure.
	E	levation of boiling point
		nd depression of
		eezing point,
		hermodynamic
		erivation of relation
	be	etween molecular
	w	eight and elevation in
		oiling point and
		epression of freezing
		oint. Experimental
	1	ethods for determining
		rious colligative
		roperties.
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Departme	_	eview the Monthly completion of Syllabus as per on plans
16 <sup>th</sup>		scheduled chapters and topics as shown in the lesson
February,	_	plan
2022		
Departme	_	eview the Monthly completion of Syllabus as per
23 <sup>rd</sup> March,		on plans scheduled chapters and topics as shown in the lesson
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2022		plan
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Departme	Departmental Meeting to Coordinate and Review the Monthly completion of Syllabus as per				
	lesson plans				
5 <sup>th</sup> May,	The teachers have completed the scheduled chapters and topics as shown in the lesson				
2022	plan				
Departme	Departmental Meeting to Coordinate and Review the Monthly completion of Syllabus as per				
_	lesson plans				
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<sup>\*</sup>Any of these – (i) Lecture Method; (ii) PPT; (iii) Online Sources; (iv) Group Discussion; (v) Case Studies etc.

Other Methods adopted by the teacher – Please write the specific teaching method