Lesson Plan

MCM DAV College for Women, Sector – 36A, Chandigarh Monthly Teaching Plans (1st Semester) Session: 2023-24

Name of the Teacher: Dr. Archana, Dr. Yesbinder Kaur and Dr. Manjot Kaur

Department: Chemistry

Class: B.Sc (1st Semester) Subject: Inorganic Chemistry

S.No.	Date (Monthly)		Topics to be Covered	Academic Activity Undertaken*
	From	To		
1.	26-07-2023	02-08-2023	Idea of de Broglie matter waves, Heisenberg uncertainty principle, atomic orbitals	Lecture
2.	03-08-2023	18-08-2023	Schrodinger wave equation, significance of Ψ and Ψ^2 , quantum numbers, radial and angular wave functions and probability distribution curves	Lecture and discussion
3.	19-08-2023	26-08-2023	Shapes of s, p, d orbitals, Aufbau and Pauli exclusion principle, Hund's multiplicity rule, electronic configuration of elements and ions	Lecture
4.	28-08-2023	13-09-2023	Position of elements in the periodic table, Effective nuclear charge and its calculation, Atomic and ionic radii, ionization energy, electron affinity and electronegativity	Lecture and group discussion
5.	14-09-2023	21-09-2023	Methods of determination of electronegativity, trends in periodic table and application in predicting and explaining the chemical behaviour	Lecture
6.	22-09-2023	30-09-2023	Chemical properties of the noble gases, chemistry of	Lecture

			vonon atmisting and	
			xenon, structure and	
			bonding in xenon	
			compounds, Comparative	
			study, diagonal	
			relationships, salient	
_	00.10.000	10 10 2022	features of hydrides	
7.	03-10-2023	19-10-2023	Solvation and	Lecture
			complexation tendencies	
			including their functions in	
			biosystems, introduction to	
			alkyls and aryls. Covalent	
			Bond- Valence bond theory	
			and its limitations	
8.	20-10-2023	31-10-2023	Directional characteristics	Lecture, Group
			of covalent bond, various	discussion and Seminar
			types of hybridizations and	
			shapes of simple inorganic	
			molecules and ions. BeF ₂ ,	
			BF ₃ , CH ₄ , PF ₅ , SF ₆ , IF ₇ ,	
			SnCl ₂ , XeF ₄ , BF ₄ , PF ₆ ,	
			SnCl ₆ ²⁻	
9.	01-11-2023	Till exam	VSEPR Theory to NH ₃ ,	Lecture, Group
			H ₃ O ⁺ , SF ₄ , ClF ₃ , ICl ₂ ⁻ and	discussion and Seminar
			H_2O , MO theory,	Lecture
			homonuclear elements and	
			ions and heteronuclear	
			(BO, CN, CO ⁺ , NO ⁺ , CO,	
			CN ⁻), diatomic molecules	
			Percentage ionic character	
			from dipole moment and	
			electronegativity	
			difference	
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Departm	Departmental Meeting to Coordinate and Review the Monthly completion of Syllabus as per		
lesson plans			
30 th August,	The teachers have completed the scheduled chapters and topics as shown in the lesson		
2023	plan		
Departmen	Departmental Meeting to Coordinate and Review the Monthly completion of Syllabus as per		
	lesson plans		
29 th Sept,	The teachers have completed the scheduled chapters and topics as shown in the lesson		
2023	plan		
Departmen	Departmental Meeting to Coordinate and Review the Monthly completion of Syllabus as per		
lesson plans			
31 th Oct,	The teachers have completed the scheduled chapters and topics as shown in the lesson		
2023	plan		

Departmental Meeting to Coordinate and Review the Monthly completion of Syllabus as per lesson plans		
22 th Nov, 2023	The teachers have completed the scheduled chapters and topics as shown in the lesson plan	

^{*}Any of these – (i) Lecture Method; (ii) PPT; (iii) Online Sources; (iv) Group Discussion; (v) Case Studies etc.Other Methods adopted by the teacher – Please write the specific teaching method

Lesson Plan

MCM DAV College for Women, Sector – 36A, Chandigarh Monthly Teaching Plans (2nd Semester) Session –2022-23

Name of the Teacher: Dr. Archana, Dr. Yesbinder Kaur and Dr. Manjot Kaur

Department: Chemistry

Class: B.Sc (2nd Semester) Subject: Inorganic Chemistry

S. No.		ate nthly)	Topics to be Covered	Academic Activity Undertaken*
	From	To		
1.	09.01.2024	31.01.2024	Ionic Solids – Concept of close packing, Ionic structures, (NaCl type, Zinc blende, Wurtzite, CaF ₂ and antifluorite)	Lecture, PPT, videos explaining the close packing and structure
2.	01-02-2024	18.02.2024	Radius ratio rule and coordination number, limitation of radius ratio rule, lattice defects, semiconductors, Lattice energy and Born-Haber cycle, solvation energy and solubility of ionic solids	Lecture, PPT
3.	19-02-2024	05-03-2024	Polarizing power and polarisability of ions, Fajan's rule. Metallic bond-free electron, valence bond and band theories. Weak Interactions – Hydrogen bonding, Vander Waals forces.	Lecture
4.	06-03-2024	16.03.2024	Comparative study (including diagonal relationship) of groups 13-14 elements, compounds like hydrides, oxides, oxyacids and halides of groups 13-14, hydrides of boron-diborane	Lecture
5.	18.03.2024	25.03.2024	Higher boranes, borazine, borohydrides, fullerenes, carbides, fluorocarbons.	Lecture and group discussion
6.	26.03.2024	08.04.2024	Comparative study of groups 15-17 elements, Compounds like hydrides, oxides	Lecture

7.	09.04.2024	Till exam	oxyacids and halides of groups	Lecture
			15-17, silicates (structural	
			principle), tetrasulphur	
			tetranitride. Basic properties of	
			halogens, interhalogens and	
			polyhalides	

_	ental Meeting to Coordinate and Review the Monthly completion of Syllabus as per
lesson plans	
30-01-2024	The teachers have completed the scheduled chapters and topics as shown in the lesson
	plan
Departmen	ntal Meeting to Coordinate and Review the Monthly completion of Syllabus as per
•	lesson plans
24-02-2024	The teachers have completed the scheduled chapters and topics as shown in the lesson
	plan
Departmen	ntal Meeting to Coordinate and Review the Monthly completion of Syllabus as per
_	lesson plans
28-03-2024	The teachers have completed the scheduled chapters and topics as shown in the lesson
	plan
Departmen	ntal Meeting to Coordinate and Review the Monthly completion of Syllabus as per
•	lesson plans
19-04-2024	The teachers have completed the scheduled chapters and topics as shown in the
	lesson plan
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^{*}Any of these – (i) Lecture Method; (ii) PPT; (iii) Online Sources; (iv) Group Discussion; (v) Case Studies etc. Other Methods adopted by the teacher – Please write the specific teaching method