Lesson Plan

MCM DAV College for Women, Sector – 36 A, Chandigarh

Monthly Teaching Plans (Odd Semester)

Session: 2025-26

Name of the Teacher: Dr. Sagarika Dev, Dr. Nisha Dawra

Department: Chemistry

Class: B.Sc (3rd Semester) Subject: Chemistry (CHM-DSC-3_Maj/Min-301)

S.No.	Date (Monthly)		Topics to be Covered	Academic Activity Undertaken*
	From	То		
1.	24-07-2025	31-08-2025	INORGANIC CHEMISTRY Chemistry of s-Block	Lecture
			Elements	
			Comparative study, diagonal relationships, salient features of hydrides.	
			ORGANIC CHEMISTRY	
			Alkenes and	
			Cycloalkenes Nomenclature of alkenes, methods of formation, mechanisms of dehydration of alcohols and dehydrohalogenation of alkyl halides, regioselectivity in alcohol dehydration.	

			PHYSICAL	
			<u>CHEMISTRY</u>	
			Thermochemistry:	
			Standard state, standard	
			enthalpy of formation-	
			Hess's Law of constant	
			Heat Summation and its	
			applications. Heat of	
			reaction at constant	
			pressure and at constant	
			volume, Enthalpy of	
			neutralisation	
2.	01-09-2025	30-09-2025	INORGANIC CHEMISTRY	Lecture and discussion
			Chemistry of s-Block	
			Elements	
			Solvation and	
			complexation tendencies,	
			including their function in	
			biosystems, and an	
			introduction to alkyls and	
			aryls.	
			<u>ORGANIC</u>	
			<u>CHEMISTRY</u>	
			Alkenes and	
			Cycloalkenes	
			Saytzeff's Rule, Hofmann	
			elimination, physical	
			properties and relative	
			stabilities of alkenes,	
			Chemical reactions of	
			alkenes-mechanisms	
			involved in hydrogenation,	
			electrophilic and free radical additions,	
			Markownikoff's rule,	
			hydroboration-oxidation,	
			oxymercuration-reduction.	
			Epoxidation, ozonolysis,	
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			hydration, hydroxylation and oxidation with KMnO4, Polymerisation of alkenes, Substitution at the allylic and vinylic positions of alkenes, Methods of formation, conformation and chemical reactions of cycloalkenes. PHYSICAL CHEMISTRY Thermochemistry: Bond dissociation energy and its calculation from thermochemical data, temperature dependence of enthalpy.Kirchhoff's equation.	
3.	01-10-2025	31-10-2025	INORGANIC CHEMISTRY Chemistry of Noble Gases Occurrence and uses, rationalisation of inertness of noble gases, Clathrates; preparation and properties of XeF ₂ , XeF ₄ and XeF ₆ ORGANIC CHEMISTRY Dienes and Alkynes Nomenclature and classification of dienes: Isolated, conjugated and cumulated dienes. Structure of allenes and butadiene, methods of formation, and polymerisation. Chemical	Lecture

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			reactions – 1,2 and 1,4	
			additions,	
			PHYSICAL	
			<u>CHEMISTRY</u>	
			Thermodynamics II:	
			Second Law of	
			Thermodynamics: Need	
			for the law, different	
			statements of the law,	
			Carnot cycle and its	
			efficiency, Carnot theorem.	
			Thermodynamic scale of	
			temperature, Concept of	
			Entropy: Entropy as a state	
			function, entropy as a	
			function of V & T	
4.	01-11-2025	10-11-2025	<u>INORGANIC</u>	Lecture and group
			<u>CHEMISTRY</u>	discussion
			Chemistry of Noble	
			Gases	
			Notes of the all and the second to	
			Nature of bonding in noble	
			gas compounds (Valence	
			bond treatment and MO	
			treatment for XeF ₂),	
			Molecular shapes of noble	
			gas compounds (VSEPR theory)	
			uicory)	
			<u>ORGANIC</u>	
			<u>CHEMISTRY</u>	
			Dienes and Alkynes	
			Diels-Alder reaction.	
			Nomenclature, structure	
			and bonding in alkynes.	
			Methods of	
			formation.Chemical	
			reactions of alkynes,	
			acidity of alkynes.	
			Mechanism of	
			electrophilic and	
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nucleophilic addition	
reactions, hydroboration-	
oxidation, metal-ammonia	
reductions, oxidation and	
polymerisation.	
PHYSICAL	
CHEMISTRY	
Thermodynamics II:	
entropy as a function of P	
& T, entropy change in	
physical change, Clausius	
inequality, entropy as a	
criterion of spontaneity	
and equilibrium. Entropy	
change in ideal gases and	
mixing of gases.	
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*Any of these – (i) Lecture Method; (ii) PPT; (iii) Online Sources; (iv) Group Discussion; (v) Case Studies etc.Other Methods adopted by the teacher – Please write the specific teaching method